

6.2 Checklist



| Lesson | Know | | Apply | | Extend |
|---|--|-----------|---|-----|---|
| 6.2.1 More about elements | I can state what an element is. | | I can identify an unknown element from its physical and chemical properties. | | I can justify the use of specific metals and non-metals for different applications, using data provided. |
| | I can state examples of elements. | | I can compare the properties of typical metals and non-metals. | | I can deduce the relationship between the position of an element in the periodic table and its properties. |
| | I can present some simple facts about an element. | \square | I can record observations and data on elements. | | I can use observations and data obtained to form conclusions about given elements. |
| 6.2.2 Chemical reactions of metals and non-metals | I can state that many elements react with oxygen to form oxides. | | I can use particle diagrams to represent oxidation reactions. | | I can decide whether a word equation represents an oxidation reaction. |
| | I can state what the arrow means in a word equation. | | I can describe an oxidation reaction with a word equation. | | I can interpret a word equation to name reactants and products. |
| | I can describe a difference in physical properties between typical metal and non-metal oxide | S. | I can classify the products obtained when typical metal and non-metal elements react w oxygen. | ith | I can deduce the physical or chemical changes a metal has undergone from its appearance. |
| 6.2.3 Metals and acids | I can describe what happens when metals react with acids. | \square | I can compare the reactions of different metals with dilute acids. | | I can suggest how temperature changes may be linked with differences in reactivity between metals with acid. |
| | I can state that when a metal reacts with an acid the products are a salt and hydrogen gas. | | I can predict the names of the products formed in a metal- acid reaction, and describe the reaction with a word equation or represent it with a particle diagr. | | |



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| | I can state which metals produce bubbles when reacting with acid. | I can decide which metals react more vigorously from practical observations. | | | |
| 6.2.4 Metals and oxygen | I can state the product of reactions between metals and oxygen. | I can compare the reactions of different metals with oxygen. | | I can explain the reactivity of metals according to how they react with oxygen. | |
| | I can name one metal that reacts vigorously with oxygen and one metal that does not react with oxygen. | I can describe an oxidation reaction with a word equation. | | I can justify the use of specific metals for different applications, using data provided. | |
| | I can make observations about how different metals react with oxygen. | I can rank metals in order of how vigorously they react with oxygen. | | I can deduce the physical or chemical changes a metal has undergone from its appearance. | |
| 6.2.5 Metals and water | I can state the products of the reaction between metals and water. | I can compare the reactions of different metals with oxygen. | | I can link a metal's reactions with its place in the reactivity series. | |
| | I can state whether a metal is more or less reactive than another metal. | I can use the reactivity series to predict reactions, and place an unfamiliar metal into the reactivity series based on information about its reaction. | | I can deduce a rule from data about which reactions will occur or not, based on the reactivity series. | |
| | I can write a simple method to find out how easily metals react with acids or water. | I can plan a practical to compare the reactivity of three metals, including identifying control variables and planning he to control them. | Dw | I can write a suitable fair test question and plan in detail which variables to control and how to control them. | |
| 6.2.6 Metal displacement reactions | I can state which metal is more reactive in a pair of named metals. | I can predict if a given pair of substances will react in displacement reactions. | | I can explain predictions about displacement reactions. | |

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| Lesson | Know | Apply | Extend | |
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| | I can state where different metals are found in the reactivity series. | I can use the reactivity series to explain displacement reactions. | I can devise a model to explain displacement reactions. | |
| | I can use observations from experiments to state whether or not a displacement reaction has occurred. | I can use word equations and particle diagrams to represent displacement reactions. | I can suggest the identity of unkown metals, given information about their reactions. | |