AQA Chemistry GCSE Student Checklist

C7 Energy changes

| Name Class Date |
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| Lesson | Aiming for 4 | Aiming for 6 | Aiming for 8 | |
|--|---|---|---|--|
| | I can define exothermic and endothermic reactions. | I can describe examples of exothermic and endothermic reactions. | I can explain a chemical reaction in terms of energy transfer. | |
| C7.1 Exothermic and endothermic reactions | I can state that energy is conserved in a chemical reaction. | I can explain, using observations from calorimetry, how to classify a reaction as exothermic or endothermic. | I can plan, carry out, and evaluate the errors in a calorimetry investigation. | |
| | I can safely complete a calorimetry experiment for a reaction that takes place in solution. | I can explain in detail how to carry out a calorimetry experiment. | | |
| | I can state a use of an exothermic reaction and an endothermic reaction. | I can explain how an energy change from a chemical reaction can be used. | I can suggest a chemical reaction for a specific purpose based on the energy change for the reaction. | |
| C7.2 Using energy transfers from reactions | I can write word equations for familiar reactions. | I can write balanced symbol equations for familiar reactions. | I can evaluate in detail the uses of exothermic and endothermic reactions. | |
| | I can define activation energy. | I can label activation energy on a reaction profile diagram. | I can explain why chemical reactions need activation energy to start them. | |
| C7.3 Reaction profiles | I can sketch a generic reaction profile diagram for an exothermic or endothermic reaction. | I can generate a specific reaction profile diagram for a given chemical reaction when its energy change is also supplied. | I can use the particle model to explain how a chemical reaction occurs. | |
| | | I can identify bonds broken in reactants and new bonds made in products of a reaction. | I can explain energy change in terms of the balance between bond making and bond breaking. | |

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| C7.4 Bond energy | | I can explain, using the particle model, how reactants become products in a chemical reaction. | | I can calculate the energy needed to break the reactant bonds and the energy released when the product bonds are made. | |
| calculations | | I can explain why bond breaking is endothermic and bond making is exothermic. | | I can calculate the energy change for a reaction, including the correct unit. | |
| | | I can define bond energy and identify all the bonds that break and are made in a chemical reaction. | | I can explain in terms of bond energies how a reaction is either exothermic or endothermic. | |