AQA Chemistry GCSE Student Checklist

C9 Crude oil and fuels

Name	Class	Date

Lesson	Aiming for 4	Aiming for 6	Aiming for 8	
	I can describe the composition of a crude oil.	I can describe how to separate crude oil into fractions in a school laboratory.	I can explain why fractional distillation is used to separate crude oil into fractions.	
C9.1 Hydrocarbons	I can state a definition of a hydrocarbon.	I can classify a hydrocarbon as an alkane.	I can apply a general formula to generate a molecular formula and a displayed formula for a straight-chain alkane.	
	I can state a definition of an alkane.	I can state the names and describe the first four alkanes.	I can classify and justify the classification of a chemical as an alkane.	
C9.2 Fractional distillation of oil	I can name the different fractions from crude oil.	I can describe how the trend in colour, viscosity, flammability, and boiling point changes as the length of the hydrocarbon chain changes.	I can explain in detail how fractional distillation is used to separate crude oil into fractions.	
	I can state a use for each fraction from crude oil.	I can describe how the properties of a fraction of crude oil make it appropriate for its use.	I can explain how chain length affects the properties of crude oil fractions.	
			I can make predictions about the properties of crude oil fractions from the fraction's hydrocarbon chain length.	

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	I can define complete and incomplete combustion.		I can explain the differences between complete and incomplete combustion.		I can justify the use of a given fuel over another.	
C9.3 Burning hydrocarbon fuels I can write a word equation to describe the complete combustion of a hydrocarbon. I can write a word equation to describe the incomplete combustion of a hydrocarbon.			I can write balanced symbol equations for the complete and incomplete combustion of hydrocarbons.		I can explain in detail how the production of carbon monoxide in incomplete combustion can be lethal.	
		I can explain how to test for the products of complete combustion.		I can use balanced symbol equations to calculate amounts of reactants or products in a combustion reaction.	C	
C9.4 Cracking hydrocarbons I can generate a w	I can define the process of cracking.		I can describe the process of cracking, including conditions.		I can use examples to explain the process of cracking and why it is so important to the petrochemical industry.	
	I can generate a word equation to describe cracking.		I can generate a balanced symbol equation to describe cracking.		I can explain the similarities and differences between alkanes and alkenes.	
	I can recognise and give examples of alkenes.		I can describe a chemical test to show an alkene is present.		I can explain, using balanced symbol equations, the reaction between bromine water and an alkene.	